

Research Article



Preparation, Pharmaceutical Characterization, *In-Vitro* Release Kinetics, and Antifungal Efficacy Investigation of Fluconazole Niosomal Hydrogel

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Abstract

Background: Lipid vesicular systems can enhance the penetration of antifungal drugs like fluconazole in topical applications. Niosomes, composed of non-ionic surfactants and cholesterol, are a key type of lipid vesicles.

Methods: Fluconazole (FL) niosomes were prepared using thin film hydration with varying ratios of Span®/Tween®/cholesterol. Their morphological characteristics, particle size, physical stability, encapsulation efficiency (EE%), cumulative drug release, and kinetics were assessed. The optimal formulation was then combined with a gel base, and its physicochemical and pharmaceutical properties were examined. Antifungal efficacy against *Candida albicans* (ATCC: 10231) was evaluated compared to free drug solutions.

Results: All formulations exhibited encapsulation efficiencies over 50%, with the Span60/Tween60/cholesterol blend (45/45/10 mole%) achieving the highest efficiency (70.2%). Following the Higuchi model, this formulation released 55.4% of FL over four hours. The gel formulations showed good physical stability, particularly the one with 1% carboxymethyl cellulose, which was suitable for topical application due to its pseudoplastic and thixotropic properties. *In-vitro* minimum inhibitory concentration (MIC) values against *Candida albicans* were recorded as 16 µg/mL (solution), 2 µg/mL (niosomal suspension), and 4 µg/mL (niosomal gel).

Conclusion: A stable and locally applicable FL niosomal gel can be formulated, potentially enhancing effectiveness and reducing microbial resistance to FL as indicated by antifungal activity results *in-vitro*.

Introduction

Fluconazole (FL) treats localized and systemic fungal infections with acceptable *in vivo* efficacy and pharmacokinetic properties. It has been derived from imidazole alcohol.^{1,2} FL inhibits ergosterol biosynthesis by inhibiting the fungal cytochrome P450 -dependent lanosterol C14 α -demethylase.³ Due to FL low solubility, its bioavailability is very low.⁴ There are oral dosage forms commercially available, which are largely associated with some side effects, such as abdominal pain, diarrhea, flatulence, nausea and vomiting, and taste disturbance after administration.⁵ Topical drug administration is widely used in various medical conditions due to its numerous benefits. These include bypassing the gastrointestinal tract, avoiding gastrointestinal irritation and the hepatic first-pass effect, and directly targeting

the affected area to minimize undesirable side effects.⁶ Unfortunately, the wide use of FL as first-line antifungal therapy has led to resistance in clinical isolates of *Candida* species including *Candida albicans* and *Candida* spp.² The resistance mechanisms fungi develop to allow them to survive at higher drug concentrations. As a result, new drugs or new drug delivery systems are urgently needed to overcome this problem.⁷ The topical delivery of lipid vesicles provides a reliable method to administer drugs directly to the infection site, reducing drug toxicity and minimizing side effects. Enhancing drug bioavailability, particularly for poorly soluble drugs, and reducing dosage and frequency while improving patient adherence lowers the overall treatment cost.⁸

Niosomes, which are vesicular nanocarriers, have garnered significant attention for their potential as drug

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delivery systems due to their distinct advantages.⁹ Several routes are available for delivering niosomes, including oral, intranasal, parenteral, and topical, as well as powders, suspensions, and semisolids.¹⁰ It enhances the permeability of drugs through the skin when applied topically and enhances the oral bioavailability of poorly soluble drugs.^{11,12} In comparison to oily dosage forms such as ointments, and creams, aqueous suspension formulations result in better patient compliance; additionally, since niosomal dispersion is aqueous, it can be emulsified in a nonaqueous phase to regulate the drug release rate.^{13,14} Amphotericin B, clotrimazole, griseofulvin, ketoconazole, and itraconazole as antifungal agents are carried by niosomes.¹⁵ These developments aimed to increase drug bioavailability while minimizing adverse effects of antifungal drugs.¹⁶ The use of a mixture of surfactants (Span[®] and Tween[®]) has increased stability and percentage of entrapment, so using a mixture of two surfactants can be a more suitable option than one surfactant.^{17,18} The semi-solid consistency achieved through gelling agents, which are mostly cellulose derivatives and natural gums at low concentrations, reduces the formulation's clearance rate and expands the residence period at the site of administration.¹⁹

In this research, we prepared various FL niosomal formulations, a mixture of Span[®] and Tween[®], and evaluated them morphologically and physicochemically. Different kinetic models were assessed to determine the best *in-vitro* kinetic model for releasing FL from niosomes. Then, the optimal niosomes formulation incorporated in the gel base and its physicochemical properties were evaluated. Finally, *in-vitro*, the antifungal effect of selected niosomal gel formulation against *Candida albicans* (ATCC:10231) was studied.

Materials and Methods

Materials

FL was a kind gift from Dr Kiafar (Zahravi Pharmaceutical Co., Tabriz, Iran). Span[®] 20, 40, and 60 and Tween[®] 20, 40, and 60 as surfactants from Fluka Company (Switzerland), and cholesterol was obtained from Sigma-Aldrich Company (USA). Culture medium, polymers, all organic solvents, and chemicals were purchased from Merck Chemical Company (Germany).

Preparation of FL niosomes

The thin-layer film hydration method was used to prepare FL niosomes.²⁰ Various niosomal formulations containing FL, surfactants, and cholesterol were prepared (The total concentration of the lipid phase in all formulations was 60 μ M presented in Table S1 (Supplementary file 1). As the organic solvent, chloroform was used to dissolve FL (final concentration: 5 mg/mL), surfactants, and cholesterol. Rotating evaporators (Heidolph, Germany) were used to remove chloroform. After the lipid film was dried, distilled water was added and rotated for 30 min at 180 rpm and 60 °C. For complete hydration, the formed lipid vesicles

Table 1. The percentage of components of FL niosomal gel formulations

Formulation name	Gelling agent	Gelling agent (%)	Propylene glycol (%)
FNG1	CMC	1	-
FNG2		2	-
FNG3	Carbomer 940	1	10
FNG4		2	10
FNG5	HPMC	2	-
FNG6		4	-

were stored at room temperature for 24 h in borosilicate glass vials and then in a refrigerator (2-8 °C) for future studies. Also, the separation of the drug entrapped in the niosomes and the free drug was performed using the dialysis method.

Morphologic study, size and zeta potential analysis of niosomal formulations

The niosomes were examined for shape, aggregation of vesicles, and separation of constituents as morphological characteristics by a light microscope (Leitz, HM-LUX3, Germany) equipped with a digital camera with $\times 400$ magnification. Mean volume diameters, and vesicular size distribution of vesicles were calculated by static laser light scattering (Malvern MasterSizer 2000E, UK) technique.²¹ Also, the zeta potential of the formulation was measured one week after its manufacture (Malvern Zetasizer Pro, UK). The range of dispersity (span), which can vary from almost monodisperse to highly polydisperse, was determined using the following equation:

$$Span = \frac{d_{v90} - d_{v10}}{d_{v50}}$$

In which d_{v90} , d_{v50} , and d_{v10} are cumulative 90, 50, and 10 percent undersize volume size distributions, respectively.²²

Determination of FL concentration

The quantity of FL was measured using UV spectrophotometry.²³ The FL standard solution (600 μ g/mL) was created in an 80:20 v/v ethanol and water mixture, and it was scanned using a UV spectrophotometer (UV/Visible Spectrophotometer Optizen 3220, South Korea) at a wavelength of 200 to 400 nm. Standard FL solutions (50-600 μ g/mL) were created once the maximum absorption wavelength (λ_{max}) was established, and the absorption of these solutions was measured at the λ_{max} . Using Microsoft Office Excel[®] software, the absorption versus concentration graph was created, and its equation was derived.

Encapsulation efficiency percent (EE%)

To calculate the EE% of FL the dialysis method was used.²⁴ Unencapsulated drugs were removed from the niosome suspensions using the dialysis method. To do this, one milliliter of the niosome suspension was placed

into a dialysis bag (Visking tube, with a molecular weight cut off of 12 KD) and dialyzed against a mixture of ethanol and water (80:20 v/v%) for four hours.^{22,25} This process effectively removed any unbound drug, leaving only the encapsulated drug within the niosome. Drug concentration was determined spectrophotometrically at 260 nm. The percentage of the drug encapsulated was calculated using the following equation:

$$EE\% = \frac{C_{\text{encap}}}{C_{\text{encap}} + C_f} \times 100$$

where C_{encap} and C_f denote the amount of FL encapsulated in niosomes and the free amount of FL, respectively.

In vitro release kinetic of FL from niosomes

Franz diffusion cell (Ashk-Shishe, Iran, Receptor volume:15 mL) and cellulose acetate dialysis tube (Visking tube, MW cut off 12 KD) as the membrane was used to study the release.²⁶ The receptor phase is a mixture of ethanol and water (80:20 v/v%) and sampling is done at different times. After determining the concentration, the graph of the cumulative released percentage of FL is drawn against time. To determine the release kinetics of FL for the optimal formulation, the *in-vitro* drug release data were fitted using a data-fitting solver in Excel software (Microsoft Office, 2016) with different release kinetic models and evaluated to understand the drug release kinetics. These included a zero-order, a first-order, Higuchi, the Korsmeyer-Peppas, and Hixon Crowell's models.²⁷

Preparation of niosomal gels

In this research, Carbomer 940⁺, carboxymethyl cellulose (CMC), and hydroxypropyl methylcellulose (HPMC) were used as gelling agents in different percentages (Table 1). In formulations containing Carbomer 940⁺, 3–5 drops of triethanolamine will be added to prepare a clear gel. Different amounts of polymer were dispersed in water and with the help of a stirrer with a constant speed of 2000 rpm, stirring continued for 30 minutes until uniformity at

25 °C, then the niosome suspension was added and mixed well. The obtained FL niosomal gel was 0.25%.

Physicochemical evaluation of niosomal Gels

Physical appearance

The prepared formulations were checked visually regarding appearance characteristics including transparency, color, absence of air bubbles, and uniformity.²⁸

Physicochemical stability

To check the physical stability in various temperatures, 15 g of each formulation was placed at -8°C for 48 h and then at 25°C for 48 h. This process was repeated for 6 cycles and the results were evaluated and reported based on the amount of leakiness, wrinkling, liquid seepage, color change, and bubble creation.²⁹ Also, to evaluate the chemical stability of FL, the selected formulation was stored at four temperatures (-8, 4, 25, and 40 °C) and the amount of FL was determined by UV spectrophotometry at one week, one month, three, and six months.³⁰

pH of formulation

One g of the prepared formulation was dispersed in 15 ml of water. pH was measured three times and reported as an average. This measurement was done at zero, 48 h, 1 week, 1 month, and 6 months after preparation of formulation (stored at room temperature).³¹

Drug content

To check the content uniformity of the gel formulation, one g of the formulation was weighed and dispersed in 5 ml of water, and then 20 ml of ethanol was added to it and the amount of FL was calculated.³¹

Rheological behaviors

The rheological behaviors test was done with a Brookfield rheometer (Brookfield Engineering, USA). The gel formulation was poured into the beaker and using the LV4 spindle at room temperature, the spindle started rotating at different speeds (at first increasing and then decreasing) and the opposite torque force was measured

Table 2. Composition, and size of different niosomal formulations containing FL (5 mg/mL)

Name	Constituents of the lipid phase	Molar ratio	Volumetric diameter (µm)			span	Zeta potential (mv)	EE%
			$d_{v10\%}$	$d_{v50\%}$	$d_{v90\%}$			
F 1	Span20/Tween20/Cholesterol	25/25/50	0.623	4.073	13.111	3.07	-15.11	ND
F 2	Span20/Tween20/Cholesterol	35/35/30	0.384	2.339	8.527	3.48	-18.22	63.7
F 3	Span20/Tween20/Cholesterol	45/45/10	0.154	0.303	0.717	1.85	-22.17	53.9
F 4	Span40/Tween40/Cholesterol	25/25/50	0.762	3.067	20.657	6.48	-19.33	ND
F 5	Span40/Tween40/Cholesterol	35/35/30	0.741	1.836	8.215	1.07	-21.22	ND
F 6	Span40/Tween40/Cholesterol	45/45/10	0.178	0.615	1.858	2.73	-29.88	66.3
F 7	Span60/Tween60/Cholesterol	25/25/50	1.122	10.081	40.103	3.87	-48.33	ND
F 8	Span60/Tween60/Cholesterol	35/35/30	1.778	3.566	10.982	2.59	-45.11	ND
F 9	Span60/Tween60/Cholesterol	45/45/10	1.435	2.770	5.062	1.31	-36.22	70.2

ND: Not Determined

at each speed. Then, a graph of revolutions per minute was drawn against the opposite torque force.³²

In-vitro FL niosomal gel release

The release rate of FL from niosomal gel was performed using a Franz diffusion cell and the same method as described in section 2.5.

In-vitro antifungal effects study

Antifungal activity in the laboratory environment of FL solution (as a control), empty niosome gel, FL niosome suspension, and FL niosome gel against *Candida Albicans* (ATCC:10231) was examined and the minimum inhibitory concentration (MIC) was calculated. Different amounts of the above formulations were added to tubes containing one ml of Sabouraud Dextrose Broth culture medium, and then concentrations (0 to 128 µg/mL) of FL were prepared by serial dilution method. Then, 0.2 ml of microbial suspension with turbidity equivalent to 1.5×10^8 CFU/mL was added to tubes. Also, two tubes were considered as positive and negative control. After 4 days of incubation at 37 °C, the tubes were checked visually for turbidity.³³

Statistical analysis

Statistical analysis was carried out using IBM SPSS Software (Version 26, IBM, USA). Group comparisons were performed through a one-way ANOVA, followed by Tukey's Multiple Comparison test, with the significance threshold set at $P < 0.05$.

Results

Determination of FL

Using the UV spectrophotometer, the $\lambda_{\max}$ of FL (in ethanol) was equal to 260 nm. This finding was consistent with the previous studies.³⁴ The equation of the line is:

$$\text{Absorbance} = 0.0019 \text{ Concentration} - 0.0118 \quad R^2 = 0.9989$$

Morphology and size analysis

After preparing different niosomes formulations, the shape of niosomes and the presence of crystals in them were checked using an optical microscope. All formulations produced niosomes, however some formulations such as F4 and F7 showed a significant amounts of likely crystal. Figure 1 shows the light microscope photo of F7 and F9.

Figure 2 shows the size distribution diagram of niosomal formulations. Based on the morphology of niosomes, the absence of crystals, and the vesicle size distribution diagram, formulations F2, F3, F6, and F9 were selected for further study. The range of dispersity (span) of thesis formulation is 3.48, 1.85, 2.73, and 1.31, respectively. Also, Table 2 shows the volumetric diameter (d_v), span, zeta potential and EE percent of the prepared formulations.

Encapsulation efficiency percent

In this study, the EE% of the prepared niosomal

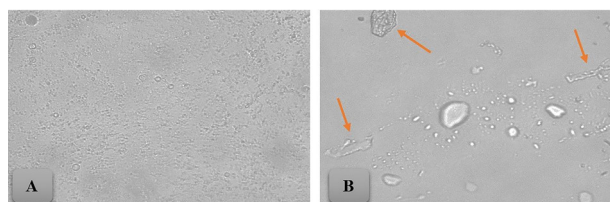


Figure 1. Optical micrographs of FL niosomes ($\times 400$ magnification): (A) F9, (B) F7 (The arrow indicates the presence of likely crystals)

formulations was above 50%, and the highest EE% was related to F 9 with a rate of 70.2. The EE% of the selected formulations is shown in Table 2.

In vitro release from niosomes

Figure 3 displays the FL release of the selected formulations. The data of the study of the release kinetics of the formulations can be found in Table 3. The highest release in niosomal formulations after 4 h is related to F6 with a value of 57.6%. Also, the release percentage in F9 is 55.4 and the release kinetic follows the Higuchi model ($R^2 = 0.975$, $k = 3.625$).

Physicochemical properties of niosomal Gels

Physical appearance

Due to the appropriate morphology, zeta potential, normal vesicle size distribution, and high EE%, the F9 formulation was chosen for further study and preparation of the niosomal gel. Formulations containing CMC had the most uniformity (FNG1, and FNG2). On the other hand, the formulation containing Carbomer 940[®] had the most air bubbles (FNG3, and FNG4). The gels containing Carbomer were transparent and light yellow (FNG3, and FNG4), while CMC and HPMC gels were transparent and colorless, tending to white (FNG1, FNG2, FNG5, and FNG6). Regarding adhesion, the gel containing 2% Carbomer (FNG3) had the highest apparent stickiness.

Physicochemical stability

To assess the shelf life of the niosomal gel formulations, they were examined after undergoing 6 thermal cycles. Most formulations were found to be stable, with no noticeable physical or organoleptic changes. However, the formulations containing Carbomer were associated with increased air bubbles, and some shrinkage was also observed in them. The formulation of selected niosomal gel (FNG1) was stable at three temperatures (-8, 4, and 25 °C) for at least six months, but at a temperature of 40 degrees after 3 months, the percentage of the remaining drug was less than 90% and instability was observed. The graph in Figure 4 displays the percentage of remaining FL at various temperatures and time intervals (Table S3.), serving as a means to evaluate the shelf life of FNG1.

pH of formulation

The formulation containing Carbomer (FNG3, and FNG4) had a higher pH than other formulations (7.71, and

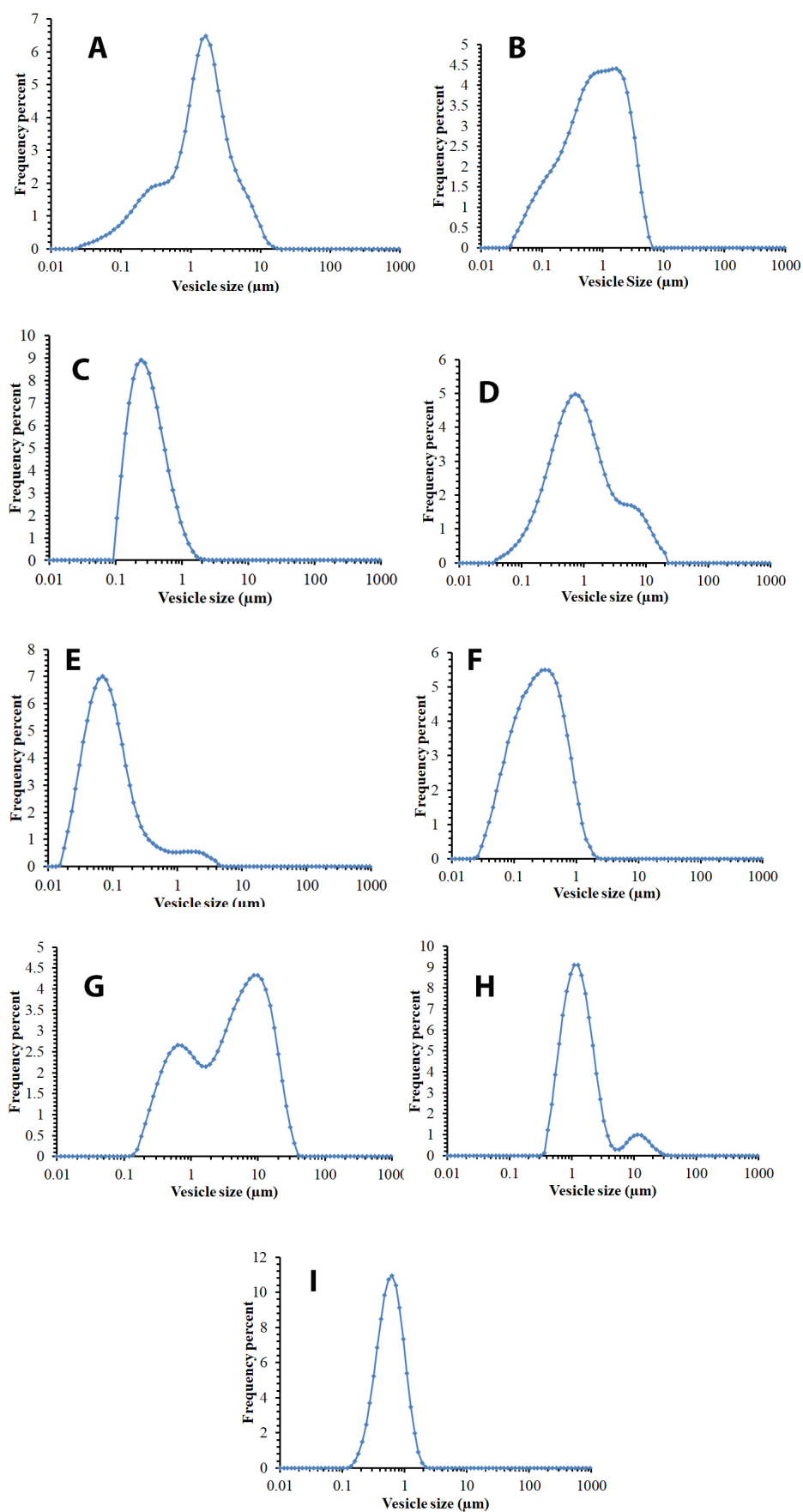


Figure 2. The vesicle size distribution diagram of niosome formulations one week after preparing and storage at 2-8°C. A: F1 (Span20/Tween20/cholesterol 25/25/50 mole%), B: F2 (Span20/Tween20/cholesterol 35/35/30 mole%), C: F3 (Span20/Tween20/cholesterol 45/45/10 mole%), D: F4 (Span40/Tween40/cholesterol 25/25/50 mole%), E: F5 (Span40/Tween40/cholesterol 35/35/30 mole%), F: F6 (Span40/Tween40/cholesterol 45/45/10 mole%), G: F7 (Span60/Tween60/cholesterol 25/25/50 mole%), H: F8 (Span60/Tween60/cholesterol 35/35/30 mole%), I: F9 (Span60/Tween60/cholesterol 45/45/10 mole%)

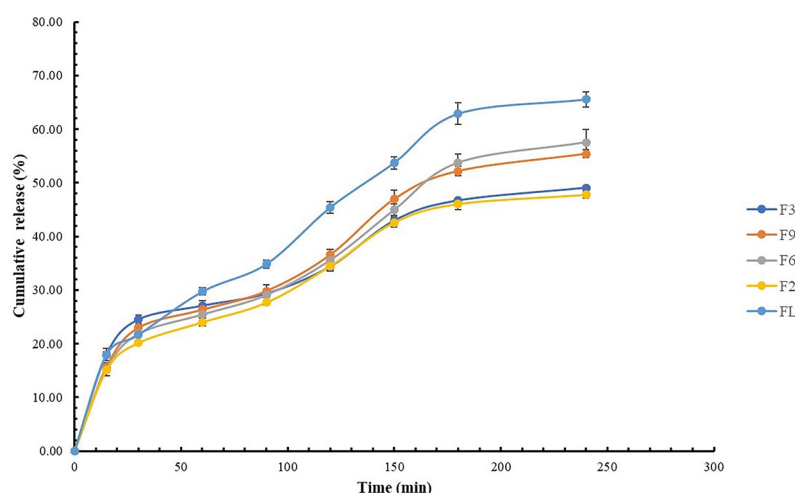


Figure 3. The release profile of FL solution (FL), and selected FL niosomal suspension (mean $\pm$ SD, $n=3$, P value=0.0367)

Table 3. Result of *in vitro* kinetic release study profile of FL niosomes and niosomal gel

Formulation	Zero-order		First order		Higuchi		Peppas		Hixon-Crowell		
	K	R ²	K	R ²	K	R ²	K	R ²	n	K	R ²
FL	0.330	0.933	0.005	0.970	4.254	0.978	0.041	0.968	0.503	0.007	0.963
F2	0.251	0.893	0.003	0.934	3.251	0.979	0.044	0.966	0.435	0.005	0.923
F3	0.258	0.853	0.003	0.911	3.367	0.965	0.067	0.945	0.358	0.004	0.894
F6	0.283	0.933	0.004	0.963	3.622	0.970	0.040	0.952	0.476	0.005	0.957
F9	0.281	0.912	0.004	0.953	3.625	0.975	0.045	0.957	0.445	0.005	0.942
FNG1*	0.190	0.892	0.002	0.925	2.545	0.990	0.023	0.984	0.514	0.003	0.915

* FNG1 is the gel formulation (Table 1) of selected FL niosome (F9).

7.02), which could be due to the use of triethanolamine to increase the pH of the environment and form a gel.^{35,36} The pH level of all formulas was suitable for topical use (5.80-7.71) and over time, small changes were observed in the form of a decrease in pH, which could be due to the absorption of CO₂ from the air and its reaction with water in the formulation.³⁷ The results of the pH measurement and its changes over six months are presented in supplementary file (Table S2).

Drug content

The content uniformity test results are presented in a supplementary file (Table S2). Based on the results, all the formulations had uniformity above 90%. Formulations containing Carbomer 940⁺ exhibited the lowest content uniformity, potentially attributed to the presence of air bubbles and its rapid gelation upon adding triethanolamine.

Rheological behaviors

The gel formulation containing Carbomer 940⁺ contained several air bubbles after adding triethanolamine and gelling. Also its drug content was less uniform and was excluded from the study. The rheological behavior of niosomal gels containing CMC and HPMC was investigated. The CMC gel showed more pseudoplastic and thixotropic behavior than the HPMC gel, making it a

more suitable candidate for topical use.^{38,39} Also, 2% CMC gel had high viscosity, and its 1% formula (FNG 1) with appropriate viscosity was selected for further study.^{40,41} Figure 5 shows the results of investigating the rheological behavior of the prepared formulations.

In vitro FL niosomal gel release

The release profile of FL niosomal gel (FNG1) is presented in Figure 6. After 4 h, the *in vitro* release of this formulation was 35.9%. Although the release rate constant of FL from niosomal gel is lower than that of niosomal suspension, it fits the Higuchi model.

In-vitro antifungal effect

Based on previous physicochemical tests, the niosomal formulation (F9) and the niosomal gel formulation (FNG1) were selected to investigate the antifungal effect. A drug-free niosomal gel formulation with similar compositions to FNG1 was prepared and studied as a control. The MIC of the prepared formulations against *Candida albicans* (ATCC: 10231) is reported in Table 4. The use of lipid bilayer drug delivery systems, such as liposome and niosome, leads to increased antifungal effects and is one of the solutions to overcome microbial resistance. In past studies, including the Musavi et al study, the use of FL liposome has increased its antifungal effects.⁴² Also, in other studies, using niosome increased the antifungal

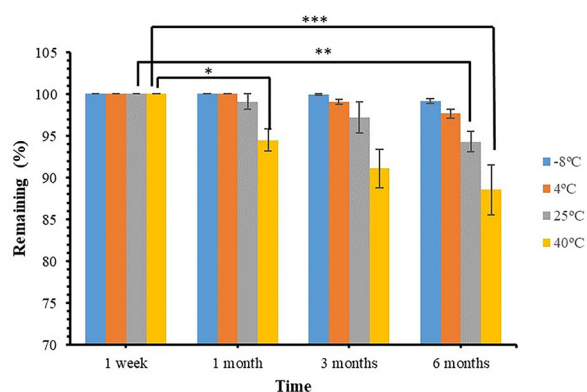


Figure 4. The remaining drug percentage at different temperatures and times for FNG1 formulation (Mean $\pm$ SD, $n=3$, * ($P<0.05$), ** ($P<0.001$), *** ($P<0.0001$)).

effects.^{15,43} The use of gel-based formulations, considering that it is more acceptable to the patient and increase the stability, can improve the effectiveness of the topical medicinal product.⁴⁴ In this study, investigating the antifungal effects of FL niosomal formulations in the laboratory environment showed a decrease in the MIC, which is consistent with previous studies.^{45,46} Niosomal gel formulation had a higher MIC than niosomal suspension, which could be due to the role of the gel base in reducing the release rate of FL.⁴⁷

Discussion

Niosomes were formed in all formulations, large amount of likely crystal was observed in some formulations, including F4 and F7. This issue can be due to the competition between cholesterol and FL as the lipophilic drug to be placed in the lipid bilayer. Past studies have used a mixture of multiple surfactants to achieve the required HLB.^{26,48,49} Cholesterol and ergosterol are also used to increase the stability of the lipid bilayer in specific proportions.^{50,51} More niosomes were observed in the formulations containing Span20/Tween20/cholesterol and Span60/Tween60/cholesterol than in the formulation containing Span40/Tween40/cholesterol. In the thin layer hydration method, niosomes are mainly formed MLV.⁵² In this study, F5, F6, and F9 were mainly MLV.

Vesicle size distribution is one of the important physicochemical properties in the preparation and optimization of niosomes, which plays a role in them *in vitro* and *in vivo* effects. Cholesterol one of the components used for different reasons such as double layer stabilizer, phase change temperature regulator, and drug release controller, can affect the vesicle size.⁵³ In the present study, it was observed that as the molar percentage of cholesterol increased, there was a corresponding increase in the size of niosomes. It has also been observed in the preparation of insulin niosomes.^{54,55} Due to the lipophilic structure of FL and structural similarity with ergosterol and cholesterol, the presence of FL can also lead to changes in vesicle size in niosomes. In the study of Shirsand et al, the presence of

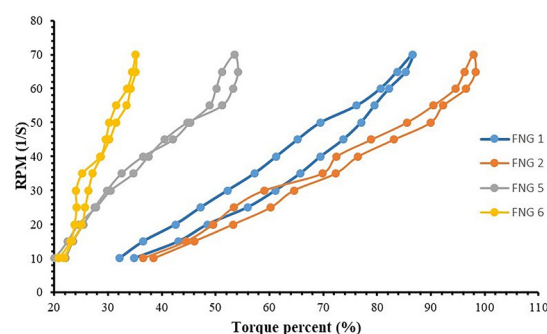


Figure 5. The results of investigating the rheological behavior of the selected niosomal gel formulations.

ketoconazole in the structure of niosomes led to a change in the vesicle size of niosomes.⁴⁷ Figure 3 illustrates that as cholesterol concentrations increase (from F9 to F7 and F6 to F4), there is a corresponding increase in the frequency of larger vesicles. This can be attributed to the higher levels of cholesterol, which leads to larger area per molecule and a thinner bilayer.⁵⁶ The observed pattern, with two distinct peaks at approximately 1 μ m and 10 μ m, which may indicate the aggregation of niosomes (F4 and F7). Formulation F9 has a bell-shaped and appropriate vesicle size distribution, while the formulations with a high percentage of cholesterol have an asymmetric vesicle size distribution. In Span20/Tween20/Cholesterol and Span40/Tween40/Cholesterol formulations, increasing cholesterol decreases negative charge density and zeta potential, as noted in prior studies.^{57,58}

FL has a lipophilic structure similar to ergosterol and cholesterol. Niosome, as a lipid bilayer structure, can potentially trap it in its bilayer. The Mousavi et al study, a similar EE% was observed for FL liposome.⁴² The results of this study show a higher EE% for FL compared to the study of Gupta et al, which can be due to our use of a mixture of two surfactants (Span[®] and Tween[®]), which leads to the formation of suitable and more stable structures.⁵⁹ In the Span20/Tween20 formulation, the EE% and particle size increased as the cholesterol percentage increased. However, due to the strong hydrophilicity (high HLB) of the Span20/Tween20 formulation, prevented a significant change in the bilayer's hydrophobicity despite the increase in cholesterol percentage. As a result, there were no significant changes in EE%. This could be due to the larger size of the niosomes, which allows for a greater amount of FL to be incorporated into the lipid bilayers.^{60,61} In general, with the increase in surfactants' hydrocarbon chain length the EE% of water-insoluble drugs such as alpha-lipoic acid²⁵, carvedilol,⁶² and flurbiprofen⁶³ increases. This study also observed this result for the surfactants Span20/Tween20, Span40/Tween40 and Span60/Tween60. As the HLB increases, the entrapment percentage of lipophilic drugs decreases. This phenomenon has also been observed in the study of curcumin niosomes. This could be attributed to the lower affinity of surfactants

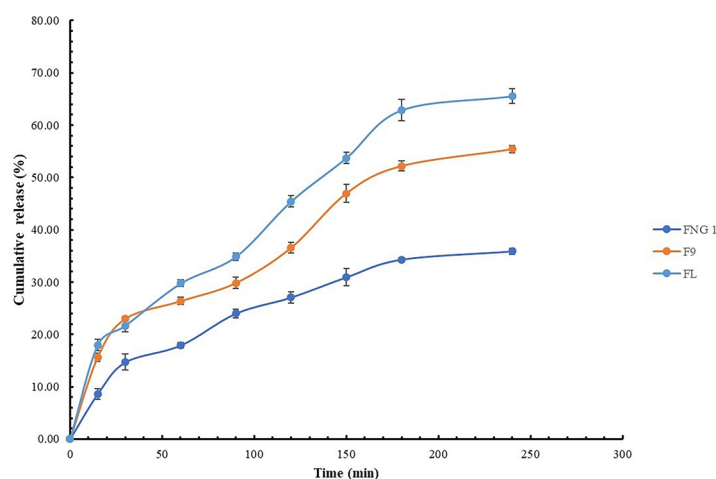


Figure 6. The release profile of FL solution (FL), FL niosomal suspension (F9), and FL niosomal gel (FNG1) (Mean $\pm$ SD, $n=3$, $P=0.0231$)

Table 4. The MIC of the prepared formulations against *Candida albicans* (ATCC: 10231)

Formulation name	Content of the formulation	MIC (μ g/mL)
FL	FL solution	16
ENG	Niosomal gel without drugs	-
F9	FL niosomal suspension	2
FNG1	FL niosomal gel	4

with higher HLB values for lipophilic molecules.⁵⁸ The EE% of the selected formulations is shown in Figure 4. The use of nonionic surfactants and encapsulation of the drug in the niosome structure leads to increased solubility of poorly soluble drugs.⁶⁰ Aldosari et al utilized various concentrations of different nonionic surfactant as stabilizers to create FL nanosuspension formulations. The impact of these stabilizers on the particle size and zeta potential of the nanosuspensions varied depending on the type and concentration of the stabilizer used. The solubility of the nanosuspension formulations improved by up to 5.7 times compared to the untreated drug.⁶⁴

In the study by El-Ridy et al, the mechanism of drug release was found to follow the Higuchi model in the niosome formulation.⁶⁵ Higuchi created various theoretical models to analyze the release of drugs soluble in water or have low solubility in semi-solid or solid forms. Mathematical equations were derived for drugs dispersed in a consistent matrix that behaves like the diffusion medium. Higuchi explains that drug release is a diffusion process dependent on the square root of time and based on Fick's law principle.²⁷ This correlation can be applied to describe the dissolution of drugs from various types of modified-release pharmaceutical dosage forms, including transdermal systems. It is important to ensure the validity of this relationship both during the initial release of the nonencapsulated drug and before the drug delivery carrier degrades and affects the diffusion coefficient. The Higuchi equation can be used to determine the apparent diffusion coefficient of a drug in a drug delivery system. This equation only has one

unknown diffusion coefficient during the release period. This coefficient can be calculated through nonlinear regression using experimental data for controlled release formulations.⁶⁶ In cases such as the release of alpha lipoic acid,²⁵ caffeine,⁶⁷ insulin,⁶⁸ and carvedilol,⁶² the drug release curve is biphasic. In this project, a biphasic release diagram was observed for niosome formulations, which could be because at first, small amounts of the drug are free or in the form of surface absorption of niosomes; It causes a release with a higher gradient, and then the FL trapped in the niosome structure is gradually released with a lower gradient from the membrane. Also, based on the Korsmeyer and Peppas models and n value the release of the FL from niosomal formulation (F9) follows Fick's law (≤ 0.45) and diffusion is the main release mechanism.²⁷ In the study conducted by El-Housiny et al, FL solid lipid nanoparticles were prepared using varying concentrations of solid lipids (Compritol 888 ATO, Precirol ATO5) and surfactants (Cremophor RH40, Poloxamer 407). The Higuchi equation best describes the drug release pattern from almost all SLN formulations, which explains the diffusion of the drug from homogeneous and granular matrix systems.⁶⁹

The release of FL from niosomal gel also follows the Higuchi model, but its release rate constant has decreased compared to niosomal suspension. This shows the role of the gel in reducing the initial release of untrapped drugs in the niosome. In the study of Moghassemi et al, this issue was also observed in the release from niosomal gel.⁵⁴ Also in Akbarzadeh et al study on the release of simvastatin⁷⁰ and Garg et al study on the release of the antifungal drug luliconazole,⁶⁰ niosomal gel formulation followed the Higuchi model.

Encapsulation of terbinafine as an antifungal agent in niosomal structure increased its effectiveness against *Aspergillus*, *Fusarium*, and *Trichophyton*.⁷¹ Niosomal formulation of antifungal drugs can be delivered transdermally instead of orally.^{72,73} Recently, Yasin et al studied, contact lenses containing FL niosomes that were prepared using Span 60 and cholesterol. The statistical

analysis revealed that contact lenses containing FL-loaded niosomes exhibited a significantly greater reduction in fungal adhesion than contact lenses containing only FL.⁷⁴ In the study conducted by Fatima et al, nano vesicular carriers were created using the ether injection method. These carriers were composed of varying levels of cholesterol, combined with free fatty acids and a monoester of polyoxyethylene fatty. The aim of the study was to investigate the antifungal effect of FL vesicles on *C. albicans*. The results showed improved efficacy and reduced MIC values compared to the FL solution.⁷⁵ In past studies, including the Musavi et al study, the use of FL liposome has increased its antifungal effects.⁴² In the study conducted by Agarwal et al, a FL niosomal gel was prepared using only Span 60 or Tween 60. The resulting niosomes were found to be spherical and their release percentage was examined. However, the study did not investigate the gel's rheological properties or the final formulation's antifungal effectiveness.⁴ The use of gel-based formulations, considering that it is more acceptable to the patient and also increase the stability, can lead to the improvement of the effectiveness of the topical medicinal product.⁴⁴ In this study, investigating the antifungal effects of FL niosomal formulations in the laboratory environment showed a decrease in the MIC, which is consistent with previous studies.^{45,46} Niosomal gel formulation had a higher MIC than niosomal suspension, which could be due to the role of the gel base in reducing the release rate of FL.⁴⁷

Conclusion

Based on the physicochemical evaluation of the prepared formulations, it is possible to create a stable and effective niosomal FL gel by using a combination of Span 60/Tween 60/cholesterol (45/45/10 mole%) as the components of the niosome and 1% CMC as a gelling agent. In laboratory testing, this formulation showed a MIC of 4 µg/mL, while the FL solution had a MIC of 16 µg/mL. This suggests that the niosomal gel could be used as a drug delivery system to enhance the effectiveness of FL. One limitation of this study is that it only examined the antifungal effect on a single strain of *Candida* and did not include any animal studies. Further studies should be conducted to determine the impact of the niosomal gel on various fungal species, such as dermatophytes. If the results are promising, this formulation could be considered for clinical trials as a topical treatment for fungal diseases in both animals and humans.

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Competing Interests

The authors declare that have no competing financial interests or personal relationships that could have appeared to influence the work reported.

Ethical Approval

In this study, the principles of ethics in research have been fully observed and approved by the Research Ethics Committees of National Institute for Medical Research Development (IR.NIMAD.REC.1399.227) and Research Ethics Committee of Kerman University of Medical Sciences (IR.KMU.REC.1401.061).

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Supplementary Files

Supplementary file 1 contains Tables S1-S3.

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