Improvement of Bioavailability and Dissolution of Tanshinone IIA by Encapsulating it with Hydroxypropyl-β-Cyclodextrin

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Abstract

Background: Tanshinone IIA (TA) could be used for the treatment of some diseases. However, the clinical application of TA was hindered by its poor water-solubility and low oral bioavailability, etc. The aim of this study was to improve its oral bioavailability and dissolution by encapsulating TA with hydroxypropyl-β-cyclodextrin (HP-β-CD).

Methods: The encapsulation composite of HP-β-CD/TA was prepared through solution method with optimization of response surface test design by taking encapsulation efficiency and drug loading as the goals; the in-vitro dissolution and in-vivo metabolism of the composite in rats were evaluated.

Results: The optimal concentration of HP-β-CD in water was 0.4 g/mL; the optimal ratio of TA to HP-β-CD was 1:7 in weight for the encapsulating process at 50°C with stirring for 2h; the encapsulation efficiency and drug loading were 84.06% and 7.38%, respectively; the cumulative release of TA from HP–β-CD TA composite in vitro dissolution reached to 72% within 90 min, which was 3.78 times of that of the original TA substance; the area under the curves of AUC(0-t) and AUC(0-∞) of the HP–β-CD -TA inclusion composite were 3.71 and 3.42 times of the original TA substance under p<0.01, respectively. The apparent distribution volume VZ/F and the clearance rate VLZ/F of HP–β-CD -TA inclusion composite decreased by 7.71 and 3.48 times as compared with the original TA substance, respectively.

Conclusion: The improvements of bioavailability and dissolution by encapsulating TA with HP-β-CD were effective.

Introduction

Tanshinone IIA (TA) is one of the main fat-soluble bioactive ingredient extracted and purified from the root of Salvia miltiorrhiza Bunge (Danshen), which has been widely used for the treatment of cardiovascular disorders, inflammation and cancer. Figure 1 showed the chemical structure of TA. However, the clinical application of TA has been hindered by its poor water-solubility (2.8 ng mL⁻¹), short half-life (1-2 h), first passes metabolism, and low oral bioavailability. Recently, numerous new drug delivery systems have been employed to deal with these problems, such as solid dispersion, polymer nanoparticle, solid liposome nanoparticle, micro-emulsions and TA sulfonate, etc. The composition of the products from these methods excluding inclusion composite is complex, which might pose unidentified hazard. Wang et al reported the preparation of inclusion composite containing TA by co-evaporating with hydroxypropyl-β-cyclodextrin (HP-β-CD) named as TA/HP-β-CD at the molar ratio of 1:1. The TA / HP-β-CD had significant dissolution rate due to the complexation with HP-β-CD. The absorption behavior of TA and the complexation through the intestinal tissues in vitro were studied without any investigation of in vivo in rats. It indicated that inclusion composite with HP-β-CD encapsulation was one of the promising approaches for improving the bioavailability of TA. HP-β-CD comprised of 6 to 8 α-D-glucopyranosides linked by α-1,4-linkages, as a result the highly water-soluble derivatives of HP-β-CD is a useful content that could be used to encapsulate...
numerous hydrophobic compounds into its cavity to improve their aqueous solubility, chemical stability and bioavailability. Some research data confirmed its nephrotoxicity based on its administration way. As the first approved CD derivative by FDA, HP-β-CD has wide applications in food, pharmaceuticals and agriculture etc. In view point of pharmacology, the pharmacokinetics of plasma metabolism differences of drug such as TA in vivo in rats was more effective to reflect its actual behavior. The aim of this study was to conduct the encapsulation for TA with HP-β-CD through the most commonly used solution method so as to increase the bioavailability of TA, and the dissolution of the composite in vitro and metabolism in vivo in rats are studied; the response surface test method with taking encapsulation efficiency and drug loading percentage as goals was used to perform the optimization of the encapsulation process of the HP-β-CD-TA inclusion composite.

Materials and Methods
TA (Purity >97%) was purchased from Xi'an Huayang Biological Technology Co. Ltd. (Shaanxi, China); the reference substance of TA (batch number: 110766-201721) was from China National Institute for the Control of Pharmaceutical and Biological Products; HP-β-CD was purchased from Xian Deli Biochemical Co., Ltd.; Diazepam Injection (2ml: 10mg) was purchased from Shanghai Xudong Haipu Pharmaceutical Co., Ltd.; Methanol (chromatographically pure) was purchased from Fisher Scientific, USA; Formic acid (analytical grade) was purchased from Tianjin Tianli Chemical Reagent Co., Ltd.; Anhydrous ethanol (analytical grade) was purchased from Tianjin Fuyu Fine Chemical Co., Ltd.; Heparin sodium injection was purchased from Shanghai First Biochemical Pharmaceuticals.

Preparation of HP-β-CD-TA inclusion composite
Solution method has been the most commonly used technique to prepare inclusion composite, which was employed to make the HP-β-CD-TA inclusion composite in this section. The procedure for preparing HP-β-CD-TA inclusion composite was as follows: weigh about 5g of HP – β - CD and dissolve it in 15mL water under normal temperature; weigh 1g of TA powder and dissolve it in 20mL ethanol, then slowly add this solution into the HP-β-CD unsaturated liquid, and stir it with magnetic stirring for some hours at certain temperature. The magnetic stirring was continued to certain time so that the encapsulation of drug molecule was sufficient, and then the unencapsulated drug molecule was removed by washing with petroleum ether three times, the organic solvent was removed by rotary evaporation at 50 °C to obtain an aqueous solution of the clathrate. The clathrate was freeze-dried further to get a clathrate powder. The freeze-drying of the vials was conducted by lyophilization for 24 h with 40 mbar vacuum in a freeze-dryer, secondary drying at 25°C for 8 h. The freeze-dryer is product of SCIENTZ-12N, Ningbo xinzhi biotechnology co. LTD, Ningbo, China.

Analytical methodology
The concentration of TA in the testing medium was quantified by high performance liquid phase chromatographic system (HPLC) of Dalian Elite C5100 Series. The testing condition was: ZORBAX Eclipse XDB-C18 (4.6×250mm, 5um) column, 2‰ formic acid water and methanol (10:90 v/v) as mobile phase at flow rate of 1.0 ml/min, and UV detection at 280 nm. The linearity range of the method was investigated for the concentration from 0.25μg/mL to 50 μg/mL (R²=0.9997). The relative standard deviation (RSD) of intraday and interday precision for TA was below 3%. The average blank recovery rates of TA in the testing medium at three concentrations (low, medium and high) were 98.13%, 97.60%, and 99.58%, respectively. The concentration of TA in the plasma content in rats was quantified by HPLC of Dalian Elite C5100 Series as well, which was under condition of ZORBAX Eclipse XDB-C18 (4.6×250mm, 5um) column, HPLC chromatographic gradient elution conditions for 2‰ formic acid water and methanol gradient elution were shown in Table 1, which were with the procedure of mobile phase at flow rate of 1.0 ml/min and UV detection at 280 nm. The standard curve of TA in rat plasma was provided to the Editorial Office of Pharmaceutical Sciences. The fitting result of the standard curve showed that the curve is highly linear within 0.19-25μ/mL. The results were, y = 140.73x + 4.30, R² = 0.9993.

Determination of encapsulation efficiency and drug loading
The encapsulation efficiency and drug loading of the encapsulation composite were assessed by using Eq. (1) and Eq. (2).

Optimization of preparation process of HP-β-CD-TA inclusion composite
Optimal design
The preparation process of HP-β-CD-TA inclusion composite was optimized by Design-Expert 8.0.6 software based on the results of single factor experiment for inclusion composite preparation in the preliminary experiment. The concentration of the drug in the preparation was controlled at 1 mg/mL, and the rotation rate of the magnetic stir was 1000 rpm. The Box-Behken response surface test design method was employed to optimize the preparation process by taking encapsulation efficiency and drug loading of the inclusion composite as the goals. The
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Encapsulation efficiency % = \( \frac{\text{Total amount of tanshinone IIA - free tanshinone IIA in the supernatant}}{\text{Total amount of tanshinone IIA}} \) \times 100\% \hspace{1cm} \text{Eq. (1)}

Drug loading % = \( \frac{\text{Total amount of tanshinone IIA - free tanshinone IIA in the supernatant}}{\text{Total amount of tanshinone IIA + amount of carrier material}} \) \times 100\% \hspace{1cm} \text{Eq. (2)}

concentration of HP-β-CD in water, the ratio of drug to carrier material, encapsulating time and temperature were taken as fundamental factors to be optimized to obtain the optimal preparation process of HP-β-CD-TA inclusion composite.

Table 2 showed the fundamental factors and level of Box-Behken test design, and Table 3 listed the design and results for HP-β-CD-TA inclusion composite prepared by Box-Behken test design. In the design, \( X_1, X_2, X_3 \), and \( X_4 \) in Tables 2 and 3 were independent variables with their meanings shown in Table 2.

Table 2. Fundamental factors and level for Box-Behken test design.

<table>
<thead>
<tr>
<th>Fundamental factor</th>
<th>Level</th>
</tr>
</thead>
<tbody>
<tr>
<td>( X_1 ): HP-β-CD concentration (mg/mL)</td>
<td>0.3 0.4 0.5</td>
</tr>
<tr>
<td>( X_2 ): Ratio of drug to carrier material</td>
<td>1:5 1:6 1:7</td>
</tr>
<tr>
<td>( X_3 ): Encapsulating time (h)</td>
<td>1 2 3</td>
</tr>
<tr>
<td>( X_4 ): Encapsulating temperature (°C)</td>
<td>40 50 60</td>
</tr>
</tbody>
</table>

**Verification test for the prediction**

Three batches of HP-β-CD-TA inclusion composite were prepared individually according to the optimal preparation process of the response surface test design. The encapsulation efficiency and drug loading of the three batches of HP-β-CD-TA inclusion composite were tested and averaged. Furthermore, the relationships of the average values for both encapsulation efficiency and drug loading of HP-β-CD-TA inclusion composite with respect to their fundamental factors were fitted to get the corresponding fitting equations by response surface test design method. Afterward, the prediction results from the fitted equations were verified by 3 verification tests.

**Fourier-transform infrared spectroscopy (FTIR)**

Proper amount of original TA, HP-β-CD, physical mixture of HP-β-CD and TA, and the inclusion composite HP-β-CD-TA were taken individually, which were ground with KBr powder uniformly and compressed into slice sample, respectively. The FTIR analysis was performed for each slice sample with a Nicolet 5700 Fourier Transform Infrared Spectrometer (Thermo Company, USA), the range of infrared spectra test was in 500 – 4000 cm\(^{-1}\).

**The vitro dissolution of HP-β-CD-TA composite**

The test for vitro dissolution of HP-β-CD-TA composite was according to procedure of the Chinese Pharmacopoeia 2015 (General Rule 0931, Cause 2 of the Part4). 0.5% sodium dodecyl sulfate solution (900 mL) was the dissolution medium, the rotation rate was 50 rpm, and the temperature was 37±0.5°C. Take a sample of 1 mL at 5, 15, 30, 45, 60, 90, 120, 180, 240 min, respectively, and replenish the same amount of dissolution medium immediately, then filter the sample with 0.45μm microporous membrane. Finally, take 10 μL of the filtrate for HPLC detection. The measurement result was substituted into the standard curve to calculate the concentration. The cumulative dissolution rate at each sampling moment was calculated using Eq. (3). Thereafter a dissolution rate curve could be plotted.

\[
Q = 100N \frac{\sum_{i=1}^{n} C_i}{W}
\]

in which Q is the cumulative dissolution rate, \( C_n \) is the mass concentration measured in the sample (mg·L\(^{-1}\)), n is the order of sampling, here the maximum of n is 9 according to the above sampling division of test; \( V_n \) is the volume of each sample (mL), and W is the amount of the inclusion composite (mg). V is the volume of the release medium (mL).

**Study of pharmacokinetics of TA composite in rats**

The pharmacokinetics was studied to realize the differences of the plasma metabolism of TA in the inclusion composite...
and the original TA substance suspension solution in rats by intra-gastric administration comparatively. The procedure of the preparation of TA substance suspension solution was as follows: accurately weigh the appropriate amount of original TA substance and place it into a mortar, add a certain volume of 0.5% sodium carboxymethyl cellulose solution, grind it to disperse uniformly and form a 8mg/mL TA suspension solution. The procedure of the preparation of HP-β-CD-TA clathrate solution was as follows: the lyophilized powder of the optimally prepared HP-β-CD-TA encapsulation composite was dissolved in a volume of ultrapure water to make HP-β-CD-TA inclusion composite solution with the concentration of TA by 8mg/mL.

Healthy Sprague-Dawley rats were randomly divided into two groups, 6 rats in each group. The rats were fasted with normal drinking water. After 12 hours, blood 0.6mL was taken from the fundus venous plexus as a blank control. Then, the TA substance suspension solution and the HP-β-CD-TA composite solution were administered by intra-gastric administration, respectively. The dosage was 100 mg/kg (calculated from the content of TA), blood was taken from the fundus venous plexus at 0.25h, 0.5h, 0.75h, 1.0h, 1.5h, 2.0h, 2.5h, 3.0h, 4.0h, 5.0h, 6h, 7h, 10.0 h after administration. The Plasma was placed in a centrifuge tube treated with sodium heparin, and centrifuged at 4°C for 10 min at the speed of 10000 revolutions per minutes (RPM). Then, the serum supernatant was accurately transferred to 100 μL, and 20 μL (50 μg/mL) of diazepam standard solution was added as an internal standard. Diazepam was the basic internal standard material, which was stable in rats. Afterward the protein was precipitated 3 times in amount of acetonitrile by vortexing for 3 min, and then centrifuged at 10000 rpm for 10 min. Finally the supernatant was taken out. The residue was precipitated twice according to the above procedure. All the resulting supernatants were combined together in a centrifuge tube, and then dried by blowing nitrogen to get the solid sample. The solid sample in the centrifuge tube was reconstituted with 100μL of methanol to obtain a solution sample. Then the solution sample was vortexed and centrifuged at 12000 rpm for 10 min at 4°C, the supernatant was taken and filtered through a 0.22μm organic filter to get final sample solution. The final sample solution was kept in refrigerator at 4°C. Take the final sample solution (plasma sample) to conduct test and inject into liquid chromatography according to its conditions, the injection volume was 20 μL, the area of TA peak was recorded, and the content of TA in plasma was calculated.

Statistical analysis for test data and presentation All data were treated by statistical analysis. The pharmacokinetic parameters were evaluated by the non-compartmental method using DAS 3.0.0 software (Shanghai, China). These parameters included the maximum plasma concentration (C_{max}), the time T_{max} till the maximum plasma concentration (C_{max}), the half-life of elimination (t_{1/2z}), the apparent distribution volume V_{d}/F, the clearance rate VL_{d}/F, the mean residence time (MRT), and the areas under the "curve of drug vs time", i.e., AUC_{0→t} and AUC_{0→∞}. The “curve of drug vs time” was stood for the curve of TA concentration in the plasmas in vivo in normal rat vs time. One-way ANOVA was used to assess statistical significance at the levels of *P < 0.05, **P < 0.01 between the different pairs of formulations.

Results and discussion

Chromatogram of TA

Figure 2 a, b and c showed the comparison of TA chromatogram in plasma spiked sample and chromatogram of the plasma samples after intra-gastric administration. The results indicated the test was effective for the detection of the concentration of TA. The linearity range of this method for the concentration of TA was in 0.19-25 μg/mL (R² = 0.9993). The limit of quantification of TA was 0.08 μg/mL, and the detection limit was 0.05 μg/mL. The RSD of intraday and interday precision and accuracy for TA were below 3% and 4%, respectively; the RSD of extraction recovery was below 7%; the RSDs for short-term room
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temperature stability and long-term stability were below 8% and 10%, respectively; and the RSD of freeze-thaw stability was below 8%.

**Optimization using response surface test method**

Fitting and analysis with response surface test method

Design-Expert 8.0.6 software was employed to analyze the experimental results listed in the Table 3, and fitted relationships to reflect the variations of encapsulation efficiency and the drug loading with respect to the fundamental factors were obtained. The statistical analysis parameters and variance analysis of the experimental results were listed in Table 4. The statistical parameters for encapsulation efficiency and drug loading were shown in Table 5 and Table 6, respectively.

The binomial equation fitted for the encapsulation efficiency $f_e$ was:

$$f_e = +84.07 + 2.41X_1 + 0.47X_2 - 3.45X_3 - 2.60X_4 - 0.89X_2X_3 - 2.51X_1X_2 + 0.43X_3X_4 + 3.45X_4X_5 - 5.11X_5 - 2.21X_2^2 - 8.46X_4^2 - 7.64X_5^2$$

Eq. (4)

The binomial equation fitted for drug loading $f_d$ was:

$$f_d = +7.38 + 0.52X_1 + 0.20X_2 - 0.34X_3 - 0.23X_4 - 0.27X_5 - 0.10X_1X_2 - 0.35X_3X_4 + 0.23X_3X_5 + 0.21X_4X_5 + 0.568X_5X_6 - 0.96X_1^2 - 0.057X_2^2 - 0.65X_3^2 - 1.08X_4^2$$

Eq. (5)

From Table 4, the fitted correlation coefficient $R^2$ of the inclusion composite and the drug loading were 0.9714 and 0.9505, respectively, and Adj were 0.9428 and 0.9011, respectively, which indicated that the model was promised; the coefficient of variation of the indicators was relatively small, 2.16 and 4.48, respectively. It showed that the credibility and precision of the model were high; the signal – to – noise ratios of the survey indicators were 18.304 and 13.317, respectively, which was greater than 4, it indicated that the response signal of the model was stronger.

From Table 5 and Table 6, it could be seen that both models were promised in statistical significance. The missing values of the indicators in these model were 69.84 and 3.21, respectively, which were greater than 0.05, it was not significant and without statistical significance, which indicated that the model could give a proper prediction without missing factor.

From Tables 5 and 6, the following results could be seen in the viewpoint of P value: the factors that affected extremely significant the encapsulation efficiency included $X_3$, $X_4$, $X_5$, $X_7$, $X_8$, $X_9$, $X_{10}$, $X_{11}$, $X_{12}$, and $X_{13}$; as to drug loading, the factors that affected extremely significant were $X_7$, $X_8$, $X_9$, $X_{10}$, $X_{11}$, $X_{12}$, and $X_{13}$; the P values of $X_1$, $X_2$, and $X_3$ were all less than 0.0001; the most significant factors that affected the encapsulation efficiency and drug loading of the inclusion composite were these three factors: $X_1$ (concentration of HP-β-CD), $X_2$ (encapsulating time) and $X_3$ (encapsulating temperature).

Above results showed that the models could be used to accurately analyze and predict the experimental results, and the optimization for the preparation process of HP-β-CD-TA inclusion composite could be conducted.

The optimal preparation process of the inclusion composite could be obtained by deriving the maximums of Eqs (4) and (5) with respect to the fundamental factors, $X_1$, $X_2$, $X_3$, and $X_4$, respectively.

Table 4. Statistical analysis parameters of the experimental results.

<table>
<thead>
<tr>
<th>Coefficient of variation (%)</th>
<th>Signal to noise ratio</th>
<th>Correlation coefficient ($R^2$)</th>
<th>Correction decision coefficient (Adj)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Encapsulation efficiency</td>
<td>2.16</td>
<td>18.304</td>
<td>0.9714</td>
</tr>
<tr>
<td>Drug loading</td>
<td>4.48</td>
<td>13.317</td>
<td>0.9505</td>
</tr>
</tbody>
</table>

Table 5. Results of statistical analysis for encapsulation efficiency in inclusion composite.

<table>
<thead>
<tr>
<th>Source of variance</th>
<th>Sum of square</th>
<th>Degree of freedom</th>
<th>Mean square</th>
<th>F value</th>
<th>P value</th>
</tr>
</thead>
<tbody>
<tr>
<td>model</td>
<td>1223.87</td>
<td>14</td>
<td>87.42</td>
<td>33.94</td>
<td>&lt;0.0001*</td>
</tr>
<tr>
<td>$X_1$</td>
<td>69.84</td>
<td>1</td>
<td>69.84</td>
<td>27.12</td>
<td>0.001**</td>
</tr>
<tr>
<td>$X_2$</td>
<td>2.61</td>
<td>1</td>
<td>2.61</td>
<td>1.01</td>
<td>0.3309</td>
</tr>
<tr>
<td>$X_3$</td>
<td>142.90</td>
<td>1</td>
<td>142.90</td>
<td>55.48</td>
<td>&lt;0.0001*</td>
</tr>
<tr>
<td>$X_4$</td>
<td>81.33</td>
<td>1</td>
<td>81.33</td>
<td>31.58</td>
<td>&lt;0.0001*</td>
</tr>
<tr>
<td>$X_5$</td>
<td>3.19</td>
<td>1</td>
<td>3.19</td>
<td>1.24</td>
<td>0.2848</td>
</tr>
<tr>
<td>$X_{10}$</td>
<td>25.20</td>
<td>1</td>
<td>25.20</td>
<td>9.78</td>
<td>0.007**</td>
</tr>
<tr>
<td>$X_{11}$</td>
<td>0.72</td>
<td>1</td>
<td>0.72</td>
<td>0.28</td>
<td>0.6047</td>
</tr>
<tr>
<td>$X_{12}$</td>
<td>47.75</td>
<td>1</td>
<td>47.75</td>
<td>18.54</td>
<td>0.0007*</td>
</tr>
<tr>
<td>$X_{13}$</td>
<td>22.90</td>
<td>1</td>
<td>22.90</td>
<td>8.89</td>
<td>0.0099*</td>
</tr>
<tr>
<td>$X_{14}$</td>
<td>69.97</td>
<td>1</td>
<td>66.97</td>
<td>27.17</td>
<td>0.001**</td>
</tr>
<tr>
<td>$X_1^2$</td>
<td>169.08</td>
<td>1</td>
<td>169.08</td>
<td>64.64</td>
<td>&lt;0.0001*</td>
</tr>
<tr>
<td>$X_2^2$</td>
<td>31.73</td>
<td>1</td>
<td>31.73</td>
<td>12.32</td>
<td>0.0035*</td>
</tr>
<tr>
<td>$X_3^2$</td>
<td>463.75</td>
<td>1</td>
<td>663.75</td>
<td>180.05</td>
<td>&lt;0.0001*</td>
</tr>
<tr>
<td>$X_4^2$</td>
<td>387.79</td>
<td>1</td>
<td>387.79</td>
<td>147.06</td>
<td>&lt;0.0001*</td>
</tr>
<tr>
<td>Residual</td>
<td>36.06</td>
<td>14</td>
<td>2.58</td>
<td>--</td>
<td>--</td>
</tr>
<tr>
<td>Missing value</td>
<td>21.39</td>
<td>10</td>
<td>2.14</td>
<td>0.58</td>
<td>0.7773</td>
</tr>
<tr>
<td>Pure error</td>
<td>14.67</td>
<td>4</td>
<td>3.67</td>
<td>--</td>
<td>--</td>
</tr>
<tr>
<td>Total deviation</td>
<td>1259.93</td>
<td>28</td>
<td>--</td>
<td>--</td>
<td>--</td>
</tr>
</tbody>
</table>

Note: 1) P<0.05 indicates significance 2) P<0.01, indicating that it is extreme significance.
The optimized fundamental factors were obtained, i.e., the HP-β-CD concentration in water was 0.4g/mL, the ratio of drug to carrier material was 1:7, and the encapsulating time and temperature were 2h and 50°C, respectively. Substituting these optimized factors into the fitting equations, the predicted encapsulation efficiency was 84.06%, and the predicted drug loading was 7.38%, respectively. The optimized fundamental factors were obtained, i.e., the HP-β-CD concentration in water was 0.4g/mL, the ratio of drug to carrier material was 1:7, and the encapsulating time and temperature were 2h and 50°C, respectively. Substituting these optimized factors into the fitting equations, the predicted encapsulation efficiency was 84.06%, and the predicted drug loading was 7.38%, respectively.

**Verification of the predictions by experiments**

The experimental results for verifying the predictions were shown in Table 7. As could be seen from Table 7 that the averaged encapsulation efficiency and drug loading of tests were 81.79% and 7.30%, respectively, of which the deviation from the predicted value was less than 3%, it indicated that the model was reliable, and could be used to conduct prediction accurately.

**Infrared characterization**

Figure 3 a, b, c and d showed the infrared characteristic absorptions of TA, HP-β-CD, simple mixture of TA and HP-β-CD, and HP-β-CD-TA inclusion composite, respectively. As could be seen from Figure 3 a), TA had a significant C=O absorption peak at 1670.4 cm^{-1}. The spectra of pure HP-β-CD and HP-β-CD-TA inclusion composite could be seen from Figure 3 b) and Figure 3 d). The peak shape on the upper part had a high degree of similarity, so it could be inferred that the basic skeleton structure of HP-β-CD had not changed during the encapsulation. Niloy et al once studied the host - guest encapsulation of active alkaloid Trigonelline hydrochloride (TgC) and hydroxypropyl – β – cyclodextrin, the comparison of FTIR spectra of both simple physical mixture and inclusion complex was used to confirm the formation of the encapsulated complex of TgC in hydroxypropyl-β-cyclodextrin encapsulation. The disappearance of absorption band between 3084 and 3040 cm^{-1} was attributed to the insertion of the C-H from -CH_{3} and C_{6} - H from aromatic ring into the cavity of the cyclodextrin moiety,^{29} thus it confirmed the formation of the encapsulated complex of TgC in hydroxypropyl-β-cyclodextrin encapsulation. The disappearance of absorption band between 3084 and 3040 cm^{-1} was attributed to the insertion of the C-H from -CH_{3} and C_{6} - H from aromatic ring into the cavity of the cyclodextrin moiety,^{29} thus it confirmed the formation of the encapsulated complex of TgC in hydroxypropyl-β-cyclodextrin encapsulation. The disappearance of absorption band between 3084 and 3040 cm^{-1} was attributed to the insertion of the C-H from -CH_{3} and C_{6} - H from aromatic ring into the cavity of the cyclodextrin moiety,^{29} thus it confirmed the formation of the encapsulated complex of TgC in hydroxypropyl-β-cyclodextrin encapsulation. The disappearance of absorption band between 3084 and 3040 cm^{-1} was attributed to the insertion of the C-H from -CH_{3} and C_{6} - H from aromatic ring into the cavity of the cyclodextrin moiety,^{29} thus it confirmed the formation of the encapsulated complex of TgC in hydroxypropyl-β-cyclodextrin encapsulation. Similarly, here the simple physical mixture of HP – β - CD and TA, and the inclusion complex of HP-β-CD-TA were comparatively shown in Figure 3 c) and d). As could be seen from Figure 3 c), the absorption peaks in the simple physical mixture of TA and HP-β-CD were the simple superposition of the absorption peaks of TA and HP-β-CD. However, in the inclusion complex diagram of Figure 3 d), the sharp characteristic absorption peak of the carbonyl C=O at 1670 cm^{-1} of TA disappeared, which was due to the shielding effect of HP-β-CD encapsulation to TA drug molecule owing to the latter entering into the cavity of after the inclusion reaction, so
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the formation of HP-β-CD-TA inclusion complex was confirmed by the above comparative analysis.\(^\text{29}\)

**In vitro dissolution test results of HP-β-CD-TA inclusion composite**

Figure 4 indicated that the dissolution of TA by HP-β-CD encapsulation was improved greatly, and the cumulative release quantity of TA from the HP-β-CD-TA inclusion composite reached to 72% within 90 min, which is 3.78 times that of the original TA substance. This is due to the dispersing and wetting actions of HP-β-CD to TA by HP-β-CD-TA inclusion composite in water,\(^\text{26,27,37,38}\) which enhanced the apparent water - solubility of TA significantly,\(^\text{26,27,37,38}\) however the original TA has poor water - solubility (2.8 ng mL\(^{-1}\)). So the results in Figure 4 reflected that improvement of in vitro cumulative release of TA by HP-β-CD-TA inclusion composite was significant.

**Pharmacokinetics of TA in the plasmas in normal rats**

Figure 5 showed the comparison of curves of TA concentration in the plasmas in vivo in normal rat vs time (drug vs time curve) for the original TA substance and HP-β-CD-TA inclusion composite by intragastric administration.

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**Figure 3.** Comparison of infrared characteristic spectra of inclusion composite, mixture and original substances of Tanshinone IIA and hydroxypropyl-β-cyclodextrin

a) Tanshinone IIA, b) hydroxypropyl-β-cyclodextrin, c) simple mixture of hydroxypropyl β-cyclodextrin and Tanshinone IIA, d) inclusion composite of hydroxypropyl –β–cyclodextrin - Tanshinone

**Figure 4.** Improvement of in vitro cumulative release of Tanshinone IIA encapsulated by hydroxypropyl – β – cyclodextrin , n = 3.

**Figure 5.** Comparison of Tanshinone IIA concentration vs time in plasma in rats by intragastric administration of Tanshinone IIA encapsulated by hydroxypropyl – β – cyclodextrin and the original Tanshinone IIA substance with dosage of 100 mg/kg (content of Tanshinone IIA), n = 6.
It could be seen from Figure 5 that the time $T_{\text{max}}$ till the concentration peak of TA of the original TA substance group in the drug-time curve was 60 min, however the corresponding time for the inclusion composite group was 180 min; the peak concentration $C_{\text{max}}$ of the former was 0.363 mg/mL and later was 0.929 mg/mL i.e., the peak concentration $C_{\text{max}}$ of inclusion composite group was 2.6 times that of the original TA substance group. Table 8 showed the pharmacokinetic parameters for original TA substance group and HP–β–CD-TA inclusion composite in 6 rats by intragastric administration. The experimental results in Table 8 showed that the both $\text{AUC}_{(0-\infty)}$ and $\text{AUC}_{(0-t)}$ were significantly different in the inclusion composite and the original TA substance under $P<0.01$, $\text{AUC}_{(0-\infty)}$ and $\text{AUC}_{(0-t)}$ of HP–β–CD-TA clathrate were 3.71 times and 3.42 times of the original TA substance, respectively. The apparent distribution volume $V_{d/F}$ and clearance rate $V_{d/F}$ of the inclusion composite were lower than that of the original TA substance, which was 7.71 times and 3.48 times lower than that of the original TA substance.

It showed that the retention time of the TA in inclusion composite was much longer than that of the original TA substance. Moreover, the $T_{\text{max}}$ till TA concentration peak $C_{\text{max}}$ of the inclusion composite was much higher than that of the original TA substance in vivo in rats, which was attributed the much larger amount of TA absorption with HP–β–CD-TA inclusion composite in the body.

The ratio of bioavailability of TA in HP–β–CD-TA inclusion composite to that of the original TA substance in vivo in rats could be defined as the relative bioavailability of TA in HP–β–CD-TA inclusion composite by Eq. (6):

$$F(\%) = \frac{[\text{AUC}_{(0-\infty)} \text{ composite}]/[\text{AUC}_{(0-\infty)} \text{ orig.}]}{[\text{AUC}_{(0-\infty)} \text{ orig.}]} \times 100\%$$

The relative bioavailability $F$ defined by Eq. (6) could reflect the improvement of bioavailability of TA by HP–β–CD encapsulation comprehensively. The relative bioavailability of TA from HP–β–CD-TA inclusion composite in vivo in rats was 342%.

**Conclusion**

The optimum preparation process of the inclusion composite was as follows: the concentration of HP–β–CD in water was 0.4 g/mL; the ratio of TA to HP–β–CD was 1:7 in weight; the encapsulating time and temperature were 2 h and 50°C, respectively. HP–β–CD inclusion composite greatly improved the in vitro dissolution of TA, the cumulative release quantity of the inclusion composite reached to 72% within 90 min, which was 3.78 times that of the original TA substance. The relative bioavailability of TA in vivo in rats was increased to 342% in HP–β–CD-TA inclusion composite. All above results indicated the effectiveness of encapsulating TA with HP–β–CD to improve its dissolution and bioavailability.

**Ethical Issues**

The study protocol was approved by the Institutional Review Board and ethical committee of Life Sciences School of Northwest University, China. (Approval Number: NWU-AWC-20180301R ).

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**Conflict of Interests**

The authors claim that there is no conflict of interest.

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